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## Structure Reports

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4-(3,5-Dioxo-10-oxa-4-azatricyclo-[5.2.1.0<sup>2,6</sup>]decan-4-yl)-10-oxa-4-azatricyclo[5.2.1.0<sup>2,6</sup>]decane-3,5-dionePeng-Peng Wang,<sup>a,b</sup> Qiu-Yue Lin<sup>a,b\*</sup> and Fan Zhang<sup>b</sup>

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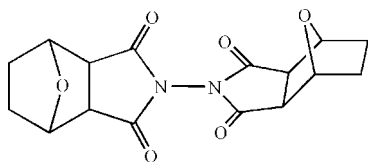
Received 30 December 2011; accepted 6 January 2012

Key indicators: single-crystal X-ray study;  $T = 296$  K; mean  $\sigma(\text{C}-\text{C}) = 0.006$  Å;  $R$  factor = 0.083;  $wR$  factor = 0.224; data-to-parameter ratio = 15.8.

In the title compound,  $\text{C}_{16}\text{H}_{16}\text{N}_2\text{O}_6$ , the dihedral angle between the two pyrrolidine rings is  $79.38$  (14)°.

## Related literature

Norcantharidin [systematic name: 7-oxabicyclo(2.2.1)heptane-2,3-dicarboxylic anhydride] and its derivatives are of significant interest as serine/threonine protein phosphatase 1 and 2A inhibitors, see: Hill *et al.* (2008). For related structures, see: Li *et al.* (2011); Zhu & Lin (2009).



## Experimental

## Crystal data

$\text{C}_{16}\text{H}_{16}\text{N}_2\text{O}_6$   
 $M_r = 332.31$   
 Orthorhombic,  $Pbca$   
 $a = 10.2342$  (6) Å  
 $b = 10.5673$  (6) Å  
 $c = 27.3485$  (17) Å  
 $V = 2957.7$  (3) Å<sup>3</sup>  
 $Z = 8$   
 Mo  $K\alpha$  radiation  
 $\mu = 0.12$  mm<sup>-1</sup>  
 $T = 296$  K  
 $0.14 \times 0.09 \times 0.08$  mm

## Data collection

Bruker P4 diffractometer  
 Absorption correction: multi-scan  
 (*SADABS*; Sheldrick, 1996)  
 $T_{\min} = 0.987$ ,  $T_{\max} = 0.991$   
 43071 measured reflections  
 3423 independent reflections  
 1581 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.163$

## Refinement

$R[F^2 > 2\sigma(F^2)] = 0.083$   
 $wR(F^2) = 0.224$   
 $S = 1.07$   
 3423 reflections  
 217 parameters  
 H-atom parameters constrained  
 $\Delta\rho_{\max} = 0.24$  e Å<sup>-3</sup>  
 $\Delta\rho_{\min} = -0.21$  e Å<sup>-3</sup>

Data collection: *SMART* (Bruker, 2004); cell refinement: *SAINTE* (Bruker, 2004); data reduction: *SAINTE*; program(s) used to solve structure: *SHELXTL* (Sheldrick, 2008); program(s) used to refine structure: *SHELXTL*; molecular graphics: *SHELXTL*; software used to prepare material for publication: *SHELXTL*.

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: FF2050).

## References

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 Li, S.-K., Zhang, F., Lv, T.-X. & Lin, Q.-Y. (2011). *Acta Cryst.* **E67**, o1974.  
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 Zhu, W.-Z. & Lin, Q.-Y. (2009). *Acta Cryst.* **E65**, o287.

**supplementary materials**

*Acta Cryst.* (2012). E68, o381 [ doi:10.1107/S1600536812000542 ]

## 4-(3,5-Dioxo-10-oxa-4-azatricyclo[5.2.1.0<sup>2,6</sup>]decan-4-yl)-10-oxa-4-azatricyclo[5.2.1.0<sup>2,6</sup>]decane-3,5-dione

P.-P. Wang, Q.-Y. Lin and F. Zhang

### Comment

Norcantharidin and its derivatives are of significant interest as serine/threonine protein phosphatase 1 and 2A inhibitors (Hill *et al.*, 2008); norcantharidin has been used in the treatment of primary hepatoma and upper gastrointestinal carcinomas, and it does not display the nephrotoxicity of cantharidin. Related norcantharidin imides were reported by Zhu & Lin (2009) and Li *et al.* (2011).

X-ray crystallography confirmed the molecular structure and the atom connectivity for the title compound, as illustrated in Fig. 1. In the molecule, the dihedral angle between the two pyrrolidine rings is 79.38 (14)°. The pyrrolidine rings are linked *via* N—N bond. The bond angles of C7—N1—N2, C8—N1—N2 and C7—N1—C8 are 121.8 (3), 123.4 (3) and 114.6 (3), respectively.

### Experimental

A mixture of 0.5 mmol norcantharidin, 0.5 mmol 2-amino-1,3,4-thiadiazole, 0.5 mmol palladium chloride as a promoter, and 10 mL distilled water was sealed in a 25 mL stainless steel reactor with a Teflon liner and heated at 393 K for 3 d. The reactor was cooled slowly to room temperature over 3 d. The solution was filtered and after 3 weeks, crystals with suitable size for single-crystal X-ray diffraction were obtained.

### Refinement

The H atoms were positioned geometrically and refined using a riding model [aliphatic of tertiary carbon C—H = 0.98 Å, aliphatic of secondary carbon C—H = 0.97 Å,  $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$ ].

### Figures

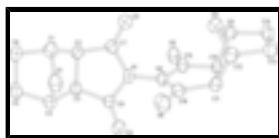


Fig. 1. A view of the molecule of the title compound showing the atom-labelling scheme with displacement ellipsoids drawn at 30% probability. Hydrogen atoms were omitted for clarity.

## 4-(3,5-Dioxo-10-oxa-4-azatricyclo[5.2.1.0<sup>2,6</sup>]decan-4-yl)-10-oxa-4-azatricyclo[5.2.1.0<sup>2,6</sup>]decane-3,5-dione

### Crystal data

C<sub>16</sub>H<sub>16</sub>N<sub>2</sub>O<sub>6</sub>

$M_r = 332.31$

$F(000) = 1392$

$D_x = 1.493 \text{ Mg m}^{-3}$

# supplementary materials

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Orthorhombic, *Pbca*  
Hall symbol: -P 2ac 2ab  
 $a = 10.2342$  (6) Å  
 $b = 10.5673$  (6) Å  
 $c = 27.3485$  (17) Å  
 $V = 2957.7$  (3) Å<sup>3</sup>  
 $Z = 8$

Mo  $K\alpha$  radiation,  $\lambda = 0.71073$  Å  
Cell parameters from 2007 reflections  
 $\theta = 1.5$ – $27.6^\circ$   
 $\mu = 0.12$  mm<sup>-1</sup>  
 $T = 296$  K  
Block, colourless  
 $0.14 \times 0.09 \times 0.08$  mm

## Data collection

Bruker P4  
diffractometer  
Radiation source: fine-focus sealed tube  
graphite  
 $\omega$  scans  
Absorption correction: multi-scan  
(*SADABS*; Sheldrick, 1996)  
 $T_{\min} = 0.987$ ,  $T_{\max} = 0.991$   
43071 measured reflections

3423 independent reflections  
1581 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.163$   
 $\theta_{\max} = 27.6^\circ$ ,  $\theta_{\min} = 1.5^\circ$   
 $h = -13 \rightarrow 13$   
 $k = -13 \rightarrow 13$   
 $l = -35 \rightarrow 35$

## Refinement

Refinement on  $F^2$   
Least-squares matrix: full  
 $R[F^2 > 2\sigma(F^2)] = 0.083$   
 $wR(F^2) = 0.224$   
 $S = 1.07$   
3423 reflections  
217 parameters  
0 restraints

Primary atom site location: structure-invariant direct methods  
Secondary atom site location: difference Fourier map  
Hydrogen site location: inferred from neighbouring sites  
H-atom parameters constrained  
 $w = 1/[\sigma^2(F_o^2) + (0.0862P)^2 + 1.8201P]$   
where  $P = (F_o^2 + 2F_c^2)/3$   
 $(\Delta/\sigma)_{\max} = 0.002$   
 $\Delta\rho_{\max} = 0.24$  e Å<sup>-3</sup>  
 $\Delta\rho_{\min} = -0.21$  e Å<sup>-3</sup>

## Special details

**Geometry.** All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

**Refinement.** Refinement of  $F^2$  against ALL reflections. The weighted  $R$ -factor  $wR$  and goodness of fit  $S$  are based on  $F^2$ , conventional  $R$ -factors  $R$  are based on  $F$ , with  $F$  set to zero for negative  $F^2$ . The threshold expression of  $F^2 > \sigma(F^2)$  is used only for calculating  $R$ -factors(gt) *etc.* and is not relevant to the choice of reflections for refinement.  $R$ -factors based on  $F^2$  are statistically about twice as large as those based on  $F$ , and  $R$ -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters ( $\text{\AA}^2$ )

	x	y	z	$U_{\text{iso}}^*/U_{\text{eq}}$
O4	0.4418 (3)	-0.1479 (3)	0.03560 (11)	0.0595 (8)
O1	0.5405 (3)	0.4020 (3)	0.13629 (11)	0.0643 (9)
N1	0.5287 (3)	0.0942 (3)	0.15182 (13)	0.0517 (9)
O2	0.3401 (3)	0.1174 (3)	0.10792 (12)	0.0658 (9)
O3	0.7107 (3)	0.1261 (3)	0.19883 (14)	0.0784 (11)
O5	0.6972 (3)	0.0800 (3)	0.07064 (12)	0.0701 (10)
C16	0.5130 (4)	-0.1318 (4)	0.14002 (15)	0.0484 (10)
C11	0.6499 (4)	-0.1471 (3)	0.06726 (14)	0.0486 (10)
H11A	0.7384	-0.1821	0.0654	0.058*
N2	0.5691 (3)	-0.0145 (3)	0.12856 (13)	0.0496 (9)
C12	0.5594 (4)	-0.2230 (3)	0.10115 (14)	0.0446 (10)
H12A	0.6021	-0.2979	0.1149	0.054*
O6	0.4401 (3)	-0.1492 (3)	0.17385 (11)	0.0654 (9)
C7	0.4108 (4)	0.1543 (4)	0.13968 (16)	0.0498 (10)
C1	0.5826 (4)	0.3963 (4)	0.18576 (17)	0.0557 (12)
H1A	0.6776	0.4021	0.1896	0.067*
C8	0.6043 (4)	0.1584 (4)	0.18598 (16)	0.0502 (11)
C10	0.4465 (4)	-0.2567 (4)	0.06668 (16)	0.0542 (11)
H10A	0.3640	-0.2742	0.0836	0.065*
C15	0.6466 (4)	-0.0144 (4)	0.08651 (16)	0.0510 (11)
C3	0.5256 (4)	0.2710 (4)	0.20309 (16)	0.0491 (10)
H3A	0.5105	0.2698	0.2385	0.059*
C2	0.4016 (5)	0.3919 (4)	0.14670 (18)	0.0622 (13)
H2A	0.3459	0.3956	0.1176	0.075*
C14	0.4899 (5)	-0.3623 (4)	0.03254 (17)	0.0622 (12)
H14A	0.4165	-0.3986	0.0150	0.075*
H14B	0.5357	-0.4286	0.0501	0.075*
C9	0.5756 (4)	-0.1560 (4)	0.01894 (16)	0.0545 (11)
H9A	0.5995	-0.0901	-0.0046	0.065*
C13	0.5819 (5)	-0.2898 (4)	-0.00196 (16)	0.0623 (13)
H13A	0.6700	-0.3235	-0.0007	0.075*
H13B	0.5507	-0.2923	-0.0354	0.075*
C4	0.3968 (4)	0.2652 (4)	0.17362 (16)	0.0500 (11)
H4A	0.3199	0.2579	0.1948	0.060*
C6	0.3774 (5)	0.4990 (4)	0.18326 (18)	0.0640 (13)
H6A	0.3051	0.4795	0.2049	0.077*
H6B	0.3602	0.5785	0.1667	0.077*
C5	0.5080 (5)	0.5027 (4)	0.21108 (18)	0.0626 (13)
H5A	0.5517	0.5835	0.2071	0.075*
H5B	0.4961	0.4856	0.2456	0.075*

Atomic displacement parameters ( $\text{\AA}^2$ )

$U^{11}$	$U^{22}$	$U^{33}$	$U^{12}$	$U^{13}$	$U^{23}$
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## supplementary materials

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O4	0.057 (2)	0.0543 (18)	0.0669 (19)	0.0109 (15)	-0.0062 (16)	-0.0045 (15)
O1	0.080 (3)	0.0474 (17)	0.065 (2)	-0.0046 (15)	0.0149 (18)	0.0010 (15)
N1	0.056 (2)	0.0350 (18)	0.064 (2)	0.0030 (16)	-0.0021 (18)	-0.0112 (16)
O2	0.058 (2)	0.0620 (19)	0.077 (2)	0.0033 (16)	-0.0110 (18)	-0.0143 (17)
O3	0.066 (2)	0.061 (2)	0.108 (3)	0.0160 (17)	-0.025 (2)	-0.0178 (19)
O5	0.074 (2)	0.0484 (17)	0.088 (2)	-0.0161 (16)	0.0124 (18)	0.0048 (17)
C16	0.057 (3)	0.041 (2)	0.046 (2)	0.004 (2)	0.002 (2)	0.001 (2)
C11	0.057 (3)	0.036 (2)	0.053 (2)	0.0058 (19)	0.004 (2)	0.0009 (18)
N2	0.056 (2)	0.0322 (17)	0.060 (2)	0.0004 (16)	0.0071 (19)	-0.0058 (16)
C12	0.050 (3)	0.0285 (19)	0.055 (2)	0.0028 (18)	0.000 (2)	0.0018 (17)
O6	0.085 (2)	0.0537 (18)	0.0576 (18)	-0.0048 (17)	0.0187 (18)	-0.0052 (15)
C7	0.047 (3)	0.042 (2)	0.061 (3)	-0.003 (2)	0.001 (2)	0.001 (2)
C1	0.051 (3)	0.039 (2)	0.078 (3)	-0.0008 (19)	-0.002 (2)	-0.003 (2)
C8	0.046 (3)	0.039 (2)	0.065 (3)	0.001 (2)	-0.005 (2)	0.000 (2)
C10	0.056 (3)	0.045 (2)	0.062 (3)	-0.005 (2)	0.005 (2)	-0.003 (2)
C15	0.047 (3)	0.043 (2)	0.063 (3)	-0.001 (2)	-0.002 (2)	0.003 (2)
C3	0.057 (3)	0.037 (2)	0.054 (2)	-0.0003 (19)	-0.001 (2)	-0.0046 (19)
C2	0.070 (3)	0.046 (2)	0.070 (3)	0.006 (2)	-0.010 (3)	0.002 (2)
C14	0.079 (3)	0.042 (2)	0.065 (3)	0.000 (2)	-0.003 (3)	-0.012 (2)
C9	0.062 (3)	0.048 (2)	0.053 (2)	0.007 (2)	0.002 (2)	0.005 (2)
C13	0.075 (3)	0.060 (3)	0.052 (3)	0.013 (2)	0.000 (2)	-0.009 (2)
C4	0.042 (2)	0.043 (2)	0.065 (3)	0.0022 (19)	0.001 (2)	-0.002 (2)
C6	0.064 (3)	0.041 (2)	0.087 (3)	0.010 (2)	-0.005 (3)	-0.004 (2)
C5	0.075 (3)	0.038 (2)	0.075 (3)	0.001 (2)	0.001 (3)	-0.011 (2)

### *Geometric parameters (Å, °)*

O4—C10	1.431 (5)	C1—H1A	0.9800
O4—C9	1.446 (5)	C8—C3	1.511 (6)
O1—C1	1.421 (5)	C10—C14	1.520 (6)
O1—C2	1.454 (6)	C10—H10A	0.9800
N1—N2	1.376 (4)	C3—C4	1.547 (6)
N1—C8	1.390 (5)	C3—H3A	0.9800
N1—C7	1.404 (5)	C2—C4	1.529 (6)
O2—C7	1.196 (5)	C2—C6	1.530 (6)
O3—C8	1.194 (5)	C2—H2A	0.9800
O5—C15	1.205 (5)	C14—C13	1.537 (6)
C16—O6	1.203 (5)	C14—H14A	0.9700
C16—N2	1.402 (5)	C14—H14B	0.9700
C16—C12	1.511 (5)	C9—C13	1.526 (6)
C11—C15	1.498 (6)	C9—H9A	0.9800
C11—C9	1.527 (6)	C13—H13A	0.9700
C11—C12	1.536 (5)	C13—H13B	0.9700
C11—H11A	0.9800	C4—H4A	0.9800
N2—C15	1.397 (5)	C6—C5	1.539 (7)
C12—C10	1.533 (6)	C6—H6A	0.9700
C12—H12A	0.9800	C6—H6B	0.9700
C7—C4	1.502 (6)	C5—H5A	0.9700
C1—C3	1.523 (6)	C5—H5B	0.9700

C1—C5	1.525 (6)		
C10—O4—C9	96.2 (3)	C1—C3—C4	101.5 (3)
C1—O1—C2	96.2 (3)	C8—C3—H3A	112.3
N2—N1—C8	123.4 (3)	C1—C3—H3A	112.3
N2—N1—C7	121.8 (3)	C4—C3—H3A	112.3
C8—N1—C7	114.6 (3)	O1—C2—C4	101.0 (3)
O6—C16—N2	124.1 (4)	O1—C2—C6	103.4 (4)
O6—C16—C12	129.7 (4)	C4—C2—C6	109.1 (4)
N2—C16—C12	106.1 (3)	O1—C2—H2A	114.0
C15—C11—C9	110.5 (3)	C4—C2—H2A	114.0
C15—C11—C12	105.3 (3)	C6—C2—H2A	114.0
C9—C11—C12	101.0 (3)	C10—C14—C13	101.0 (3)
C15—C11—H11A	113.1	C10—C14—H14A	111.6
C9—C11—H11A	113.1	C13—C14—H14A	111.6
C12—C11—H11A	113.1	C10—C14—H14B	111.6
N1—N2—C15	123.4 (3)	C13—C14—H14B	111.6
N1—N2—C16	120.8 (3)	H14A—C14—H14B	109.4
C15—N2—C16	114.6 (3)	O4—C9—C13	102.3 (3)
C16—C12—C10	110.1 (3)	O4—C9—C11	101.2 (3)
C16—C12—C11	106.3 (3)	C13—C9—C11	111.1 (3)
C10—C12—C11	101.8 (3)	O4—C9—H9A	113.7
C16—C12—H12A	112.6	C13—C9—H9A	113.7
C10—C12—H12A	112.6	C11—C9—H9A	113.7
C11—C12—H12A	112.6	C9—C13—C14	101.9 (3)
O2—C7—N1	123.0 (4)	C9—C13—H13A	111.4
O2—C7—C4	130.2 (4)	C14—C13—H13A	111.4
N1—C7—C4	106.8 (4)	C9—C13—H13B	111.4
O1—C1—C3	102.5 (3)	C14—C13—H13B	111.4
O1—C1—C5	104.4 (4)	H13A—C13—H13B	109.3
C3—C1—C5	107.9 (4)	C7—C4—C2	112.5 (4)
O1—C1—H1A	113.7	C7—C4—C3	105.7 (3)
C3—C1—H1A	113.7	C2—C4—C3	100.9 (3)
C5—C1—H1A	113.7	C7—C4—H4A	112.3
O3—C8—N1	124.5 (4)	C2—C4—H4A	112.3
O3—C8—C3	128.3 (4)	C3—C4—H4A	112.3
N1—C8—C3	107.2 (4)	C2—C6—C5	101.6 (4)
O4—C10—C14	103.6 (3)	C2—C6—H6A	111.4
O4—C10—C12	101.7 (3)	C5—C6—H6A	111.4
C14—C10—C12	109.1 (4)	C2—C6—H6B	111.4
O4—C10—H10A	113.8	C5—C6—H6B	111.4
C14—C10—H10A	113.8	H6A—C6—H6B	109.3
C12—C10—H10A	113.8	C1—C5—C6	101.1 (4)
O5—C15—N2	122.7 (4)	C1—C5—H5A	111.6
O5—C15—C11	129.7 (4)	C6—C5—H5A	111.6
N2—C15—C11	107.5 (3)	C1—C5—H5B	111.6
C8—C3—C1	112.6 (4)	C6—C5—H5B	111.6
C8—C3—C4	105.2 (3)	H5A—C5—H5B	109.4

Fig. 1

